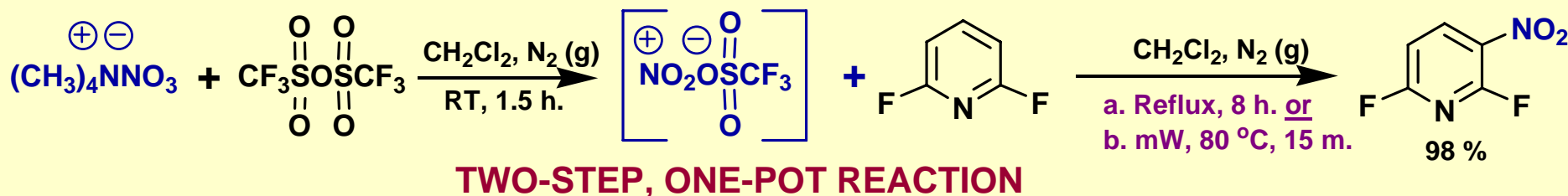




# ANHYDROUS NITRONIUM TRIFLATE NITRATION OF AROMATIC and HETEROAROMATIC COMPOUNDS: CONVENTIONAL BENCHTOP & MICROWAVE- ASSISTED CONDITIONS

Scott A. Shackelford, Wade W. Grabow, Ashwani Vij

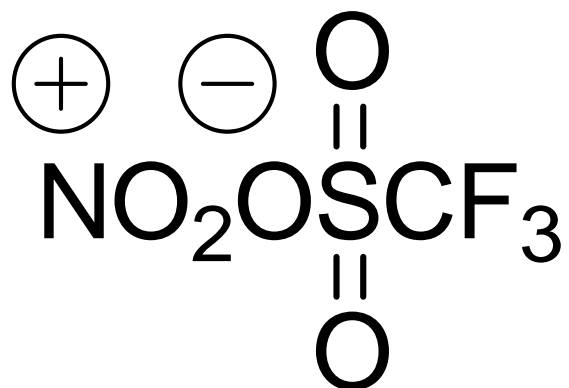
Air Force Research Laboratory  
Propellants Branch  
10 East Saturn Drive  
Edwards AFB, CA 93524-7680



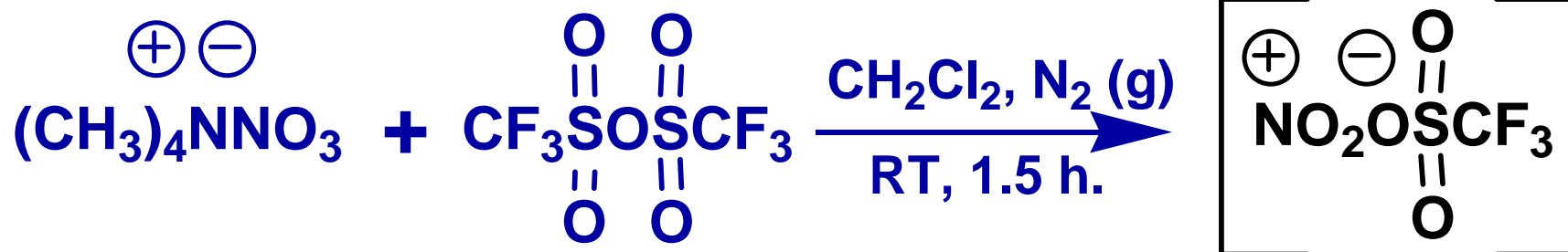
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# ANHYDROUS NITRONIUM TRIFLATE NITRATION



## CONVENIENT IN-SITU GENERATION





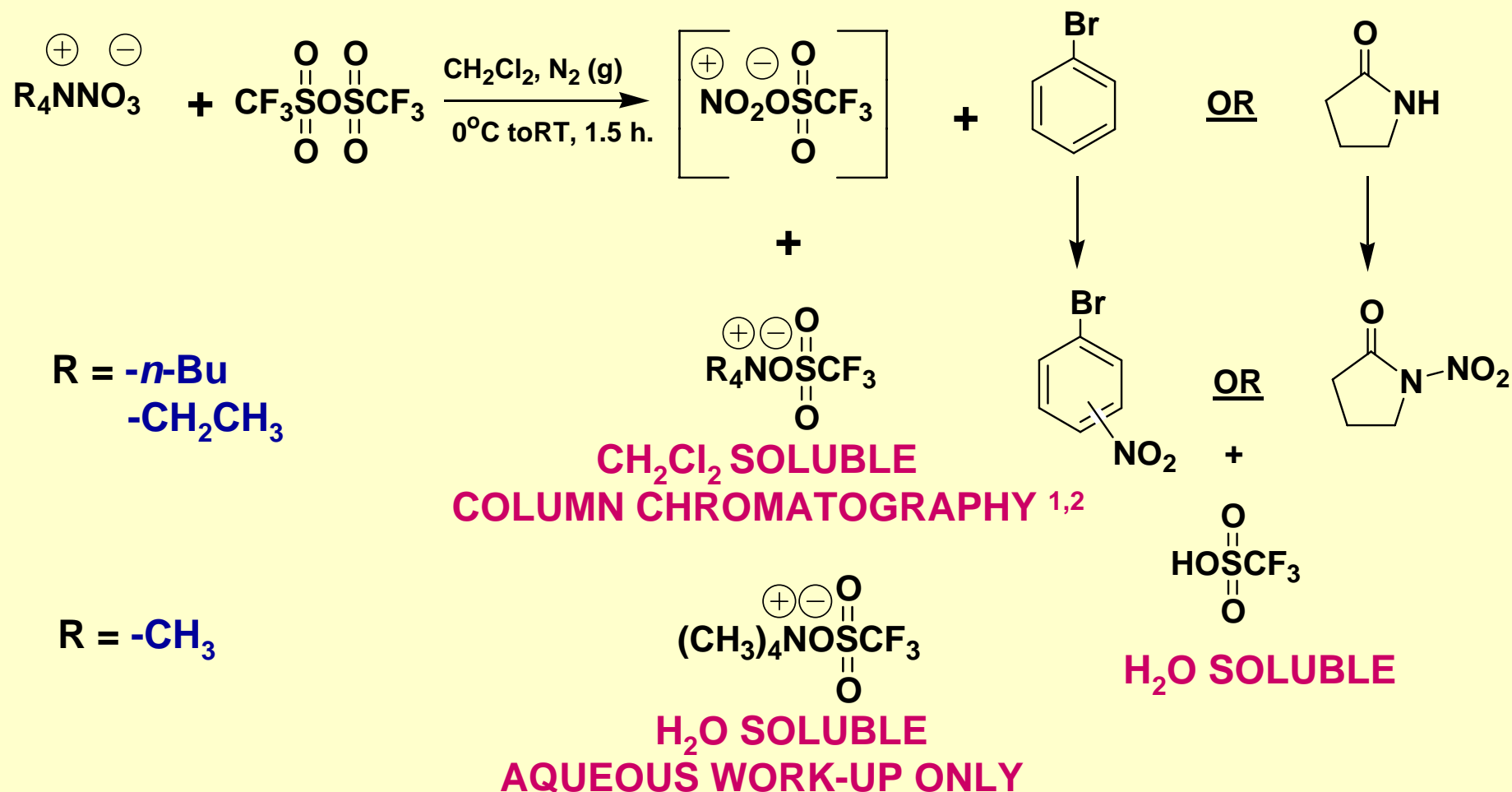
# OVERVIEW



- **IMPROVED  $\text{NO}_2\text{OSO}_2\text{CF}_3$  NITRATING REAGENT *IN-SITU* FORMATION**
  - **$(n\text{-Bu})_4\text{NNO}_3$  GENERATED REAGENT**
  - **$(\text{CH}_3)_4\text{NNO}_3$  GENERATED REAGENT**
  
- **CONVENTIONAL BENCHTOP & MICROWAVE-ASSISTED  $\text{NO}_2\text{OSO}_2\text{CF}_3$  NITRATION RESULTS**
  - **CONVENTIONAL BENCHTOP LIMITATIONS**
  - **MICROWAVE-ASSISTED EXTENTIONS**
  
- **PRELIMINARY MICROWAVE-ASSISTED  $\text{NO}_2\text{OSO}_2\text{CF}_3$  NITRATION SCREENING REACTIONS**
  - **DIRECTIONAL EFFECTS OF 5-MEMBERED HETEROAROMATICS**
  - **POSITIONAL NITRATION REACTIVITY**



# ANHYDROUS NITRONIUM TRIFLATE NITRATION



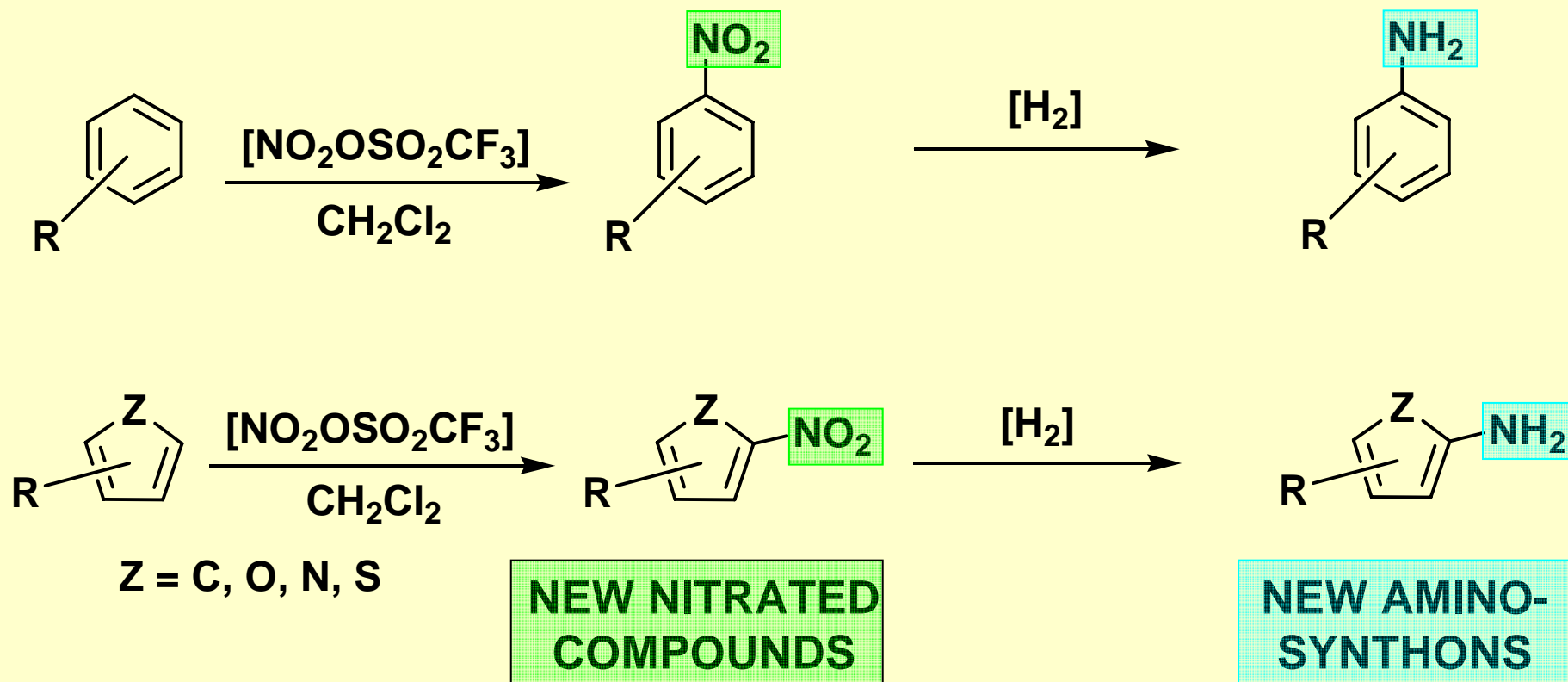
1. Shackelford, S. A.; Adams, C. M.; Sharts, C. M. 11<sup>th</sup> Rocky Mountain Region ACS Mtg., Albuquerque, NM, 10-12 Jul 1992.
2. Adams, C. M.; Sharts, C. M.; Shackelford, S. A. *Tetrahedron Lett.* **1993**, 34, 6669-6671.



# ANHYDROUS NITRONIUM TRIFLATE NITRATION

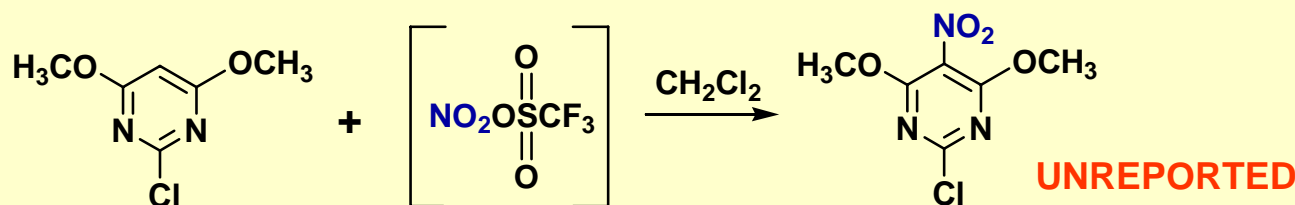


## OBJECTIVE: NEW AROMATICS & HETEROAROMATICS

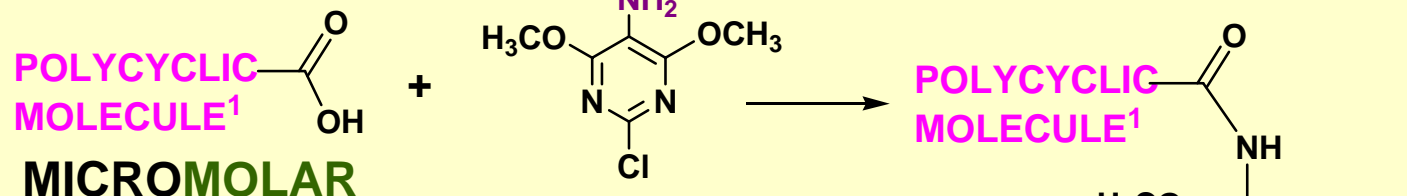




# ANHYDROUS NITRONIUM TRIFLATE NITRATION

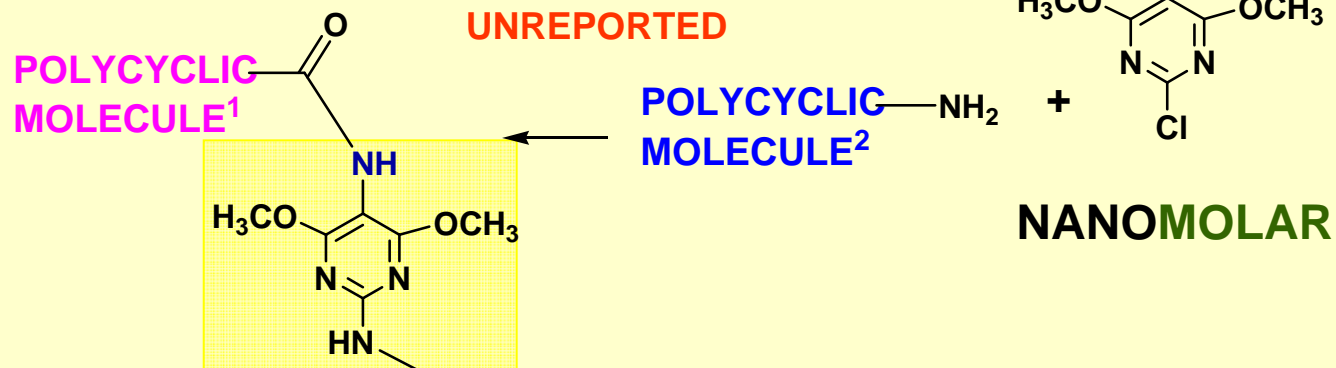


**BIOASSAY  
RESPONSES**

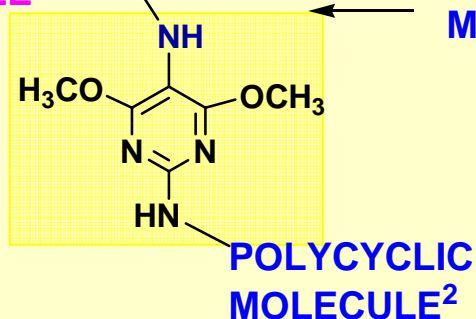


**MICROMOLAR**

**UNREPORTED**



**NANOMOLAR**



**PICOMOLAR**

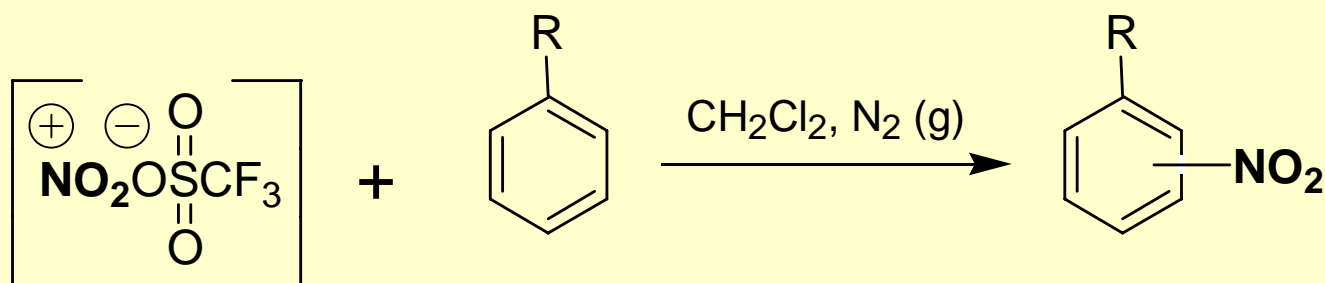
**7 PHARMACEUTICAL  
CANDIDATES**



# ANHYDROUS NITRONIUM TRIFLATE NITRATION



## ➤ NITRATION CONDITIONS:

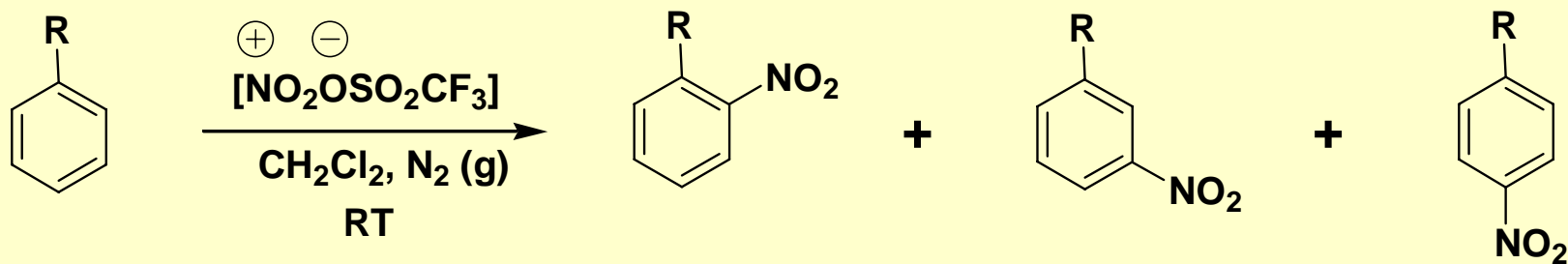


	<u>EQUIVALENTS</u>	<u>CONCENTRATION</u>	<u>TEMPERATURE</u>	<u>METHOD</u>
<u>DILUTE:</u>	1.05 to 1.7	0.2 to 0.3M	-78 °C to RT	CONVENTIONAL BENCHTOP
<u>CONC.:</u>	1.5 to 2.0	1.1 to 3.6M	RT to >REFLUX	CONVENTIONAL BENCHTOP & MICROWAVE- ASSISTED

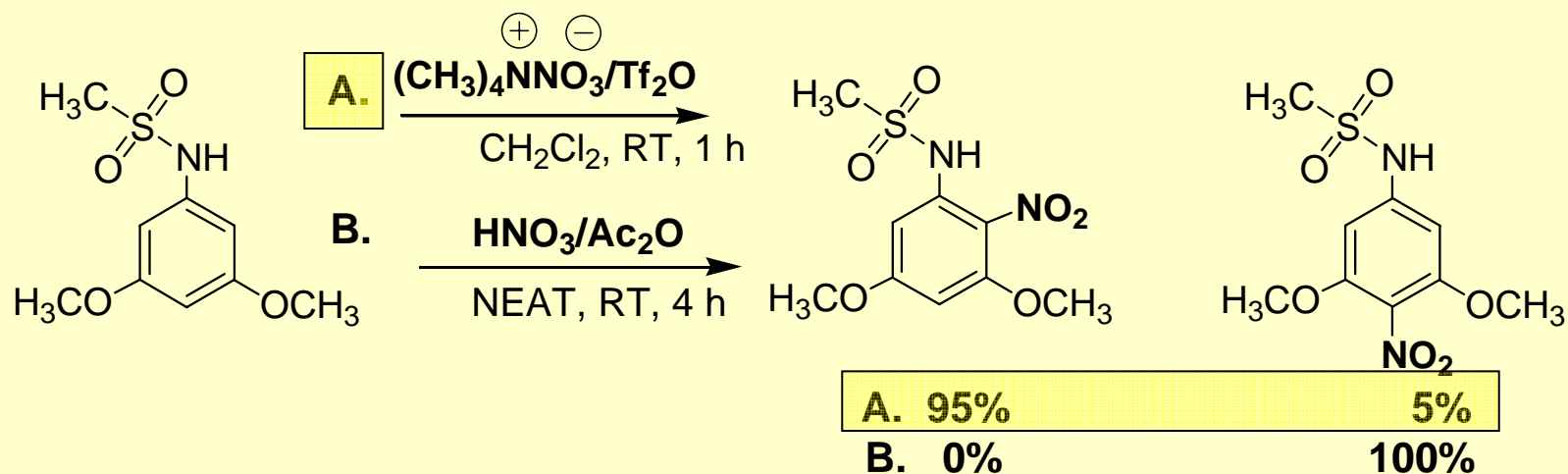




# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



R = -OCH <sub>3</sub>	63%	5%	32%
-OH	90	0	10
-CH <sub>3</sub>	62	0	38
-Br	33	0	67
-CO <sub>2</sub> H	16	78	6
-CF <sub>3</sub>	8%	88%	4%



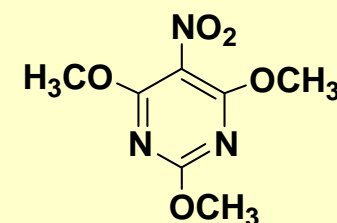
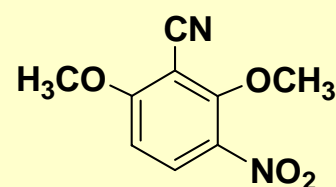
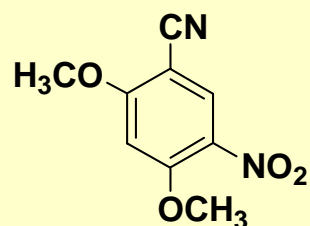
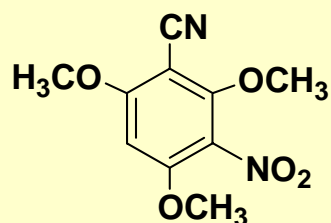
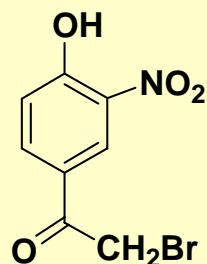


# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



## ➤ NITRATION RESULTS:

### PRODUCT:



UNREPORTED

ONLY ISOMER

DIRECT NITRATION

### SCALE

(mmol):

8 93

8 100

100

100

371

### ISOLATED

YIELD:

94% 91%

86% 82%

91%

100%

82%

### PURITY:

91% 88%

95% 95%

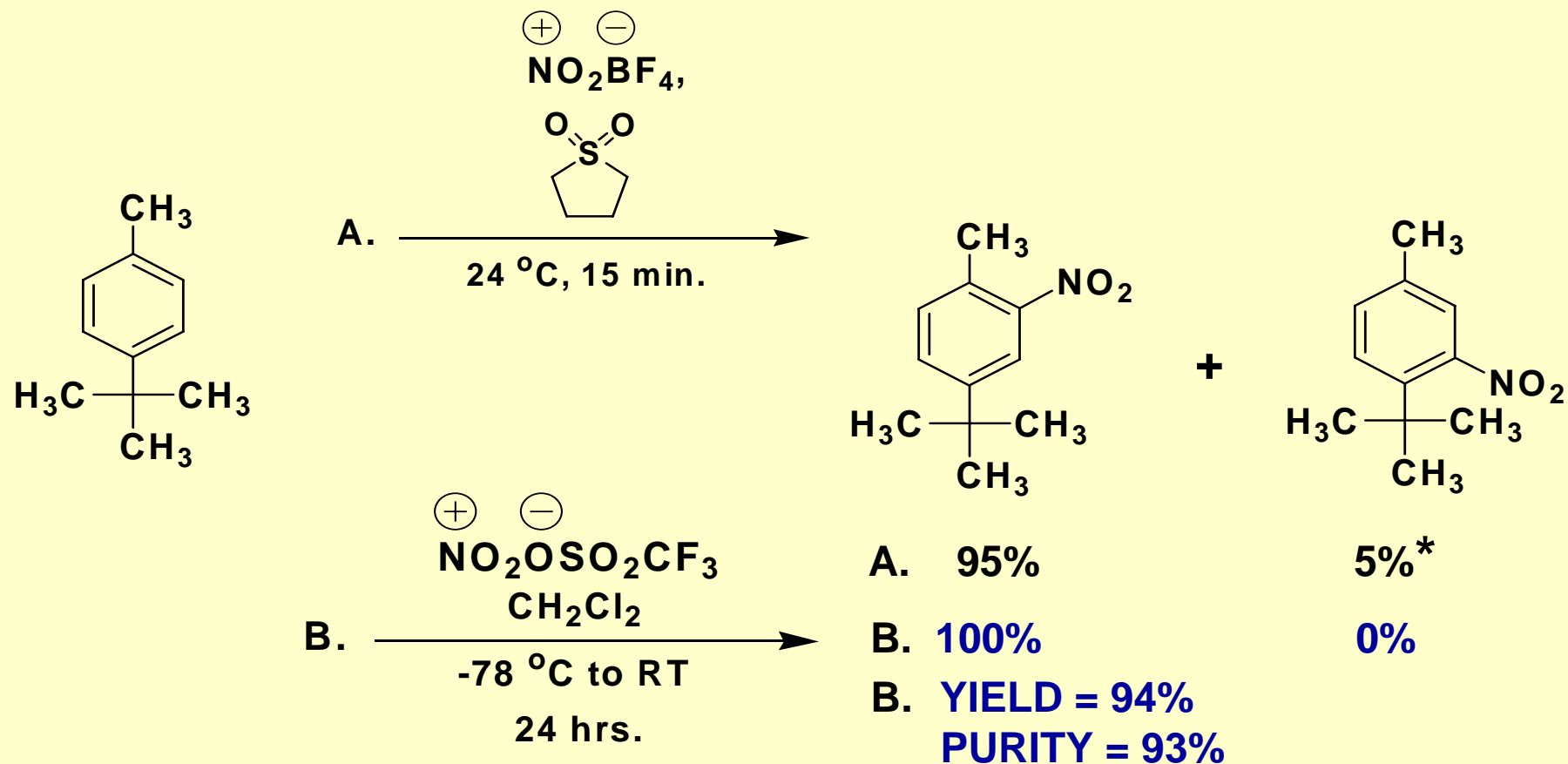
82%

97%

98%



## ➤ NITRATION SELECTIVITY (82 mmol):



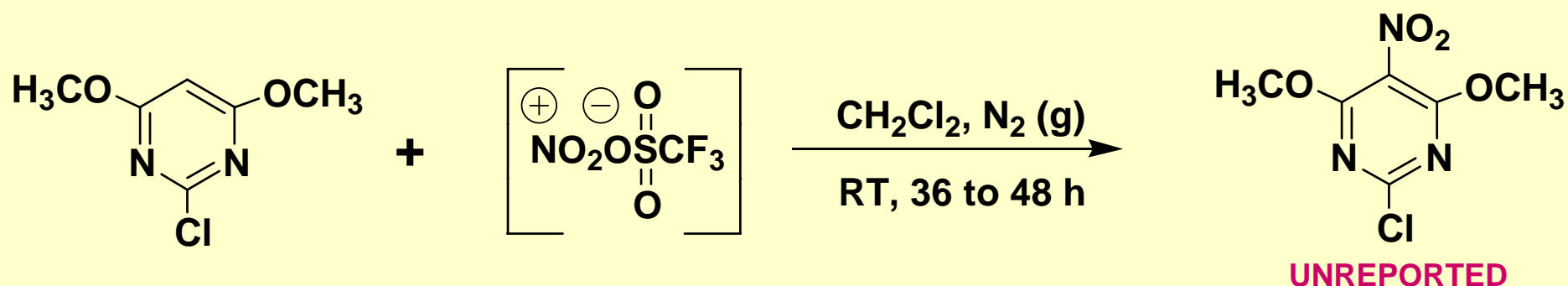
\* Olah, G. A.; Kuhn, S. *J. Am. Chem. Soc.*, **1964**, 86, 1067-1070.



# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



## ➤ $\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIVALENTS & NITRATION SCALE-UP:



<u>CONDITIONS</u>	<u>SCALE (mmol)</u>	<u><math>\text{NO}_2\text{OSO}_2\text{CF}_3</math> EQUIV.</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD.</u>	<u>PURITY</u>
[0.2M], 48 h	8	1.05	68	-----	68%
	8	1.25	88	-----	88%
	8	1.50	100	98%	99%*
[3.6M], 36 h	2864	1.5	100	94%	>95%

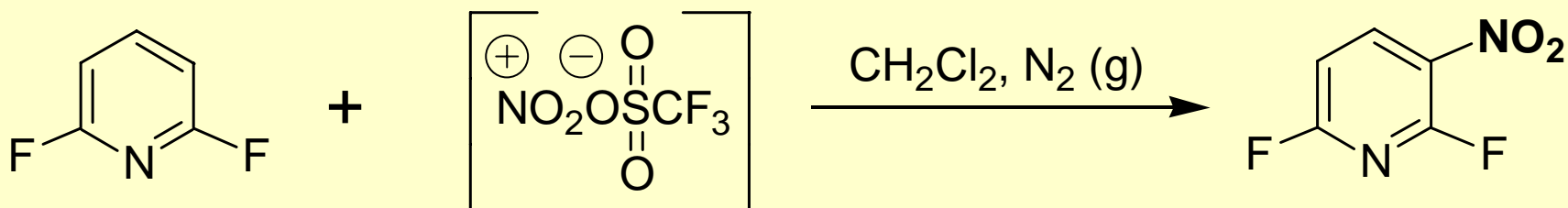
\* ANALYTICALLY PURE BY C, H, Cl, and N, ELEMENTAL ANALYSES



# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



## ➤ NITRATION REACTIVITY:



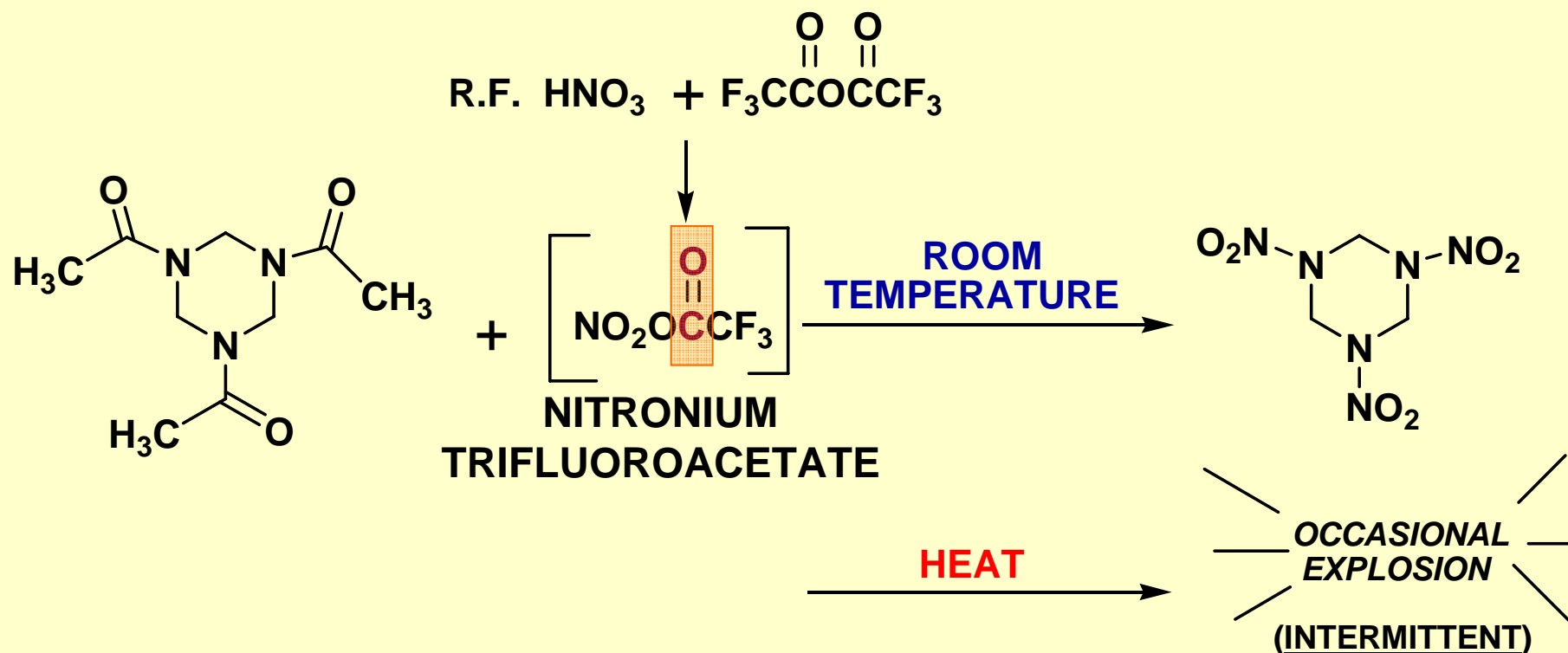
<u>CONDITIONS</u>	<u>NO<sub>2</sub>OSO<sub>2</sub>CF<sub>3</sub> EQUIV.</u>	<u>[REACTANT CONC.]</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
RT, 42 h	1.6	0.25M	69%	-----	<69%
RT, 24 h	1.5	1.09M	78%	-----	78%



# CONVENTIONAL BENCHTOP NITRONIUM TRIFLUOROACETATE NITRATION



## ➤ OTHER NITRATION REAGENT REACTIVITY:





# MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



## ➤ ORIGINAL MICROWAVE INSTRUMENTS EMPLOYED:

▪ PERSONAL MICROWAVE™

**SMALL SCALE REACTIONS**

**3.4 mmol (0.5 gram)**

▪ MILESTONE ETHOS™

**LARGER-SCALE REACTIONS**

**54.5 mmol (10 grams)**

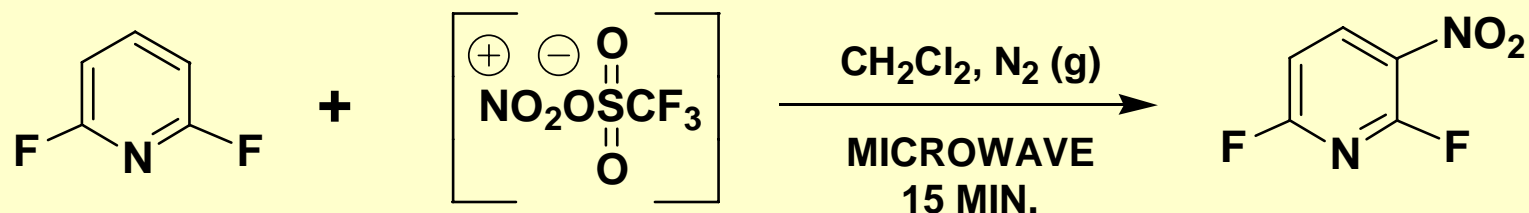




# MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



## ➤ MICROWAVE TEMPERATURE SCREENING:



<u>RUN NO.</u>	<u>NO<sub>2</sub>OSO<sub>2</sub>CF<sub>3</sub> EQUIV.</u>	<u>TEMP (°C)</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
1	1.3	60	91	-----	-----
8	1.5	60	96	85%	-----
5	1.5	110	100	91%	98%
7	1.5	100	100	91%	98%
9	1.5	80	100	94%	100%*

\*ANALYTICALLY PURE BY C, H, F, & N ELEMENTAL ANALYSIS

- INCREASED NO<sub>2</sub>OSO<sub>2</sub>CF<sub>3</sub> EQUIVALENTS IMPROVE REACTANT CONVERSION (RUNS 1 & 8)
- INCREASED TEMPERATURE RAISES REACTANT CONVERSION (RUN 5)
- MINIMUM TEMPERATURE (80 °C) EXISTS FOR COMPLETE REACTION (RUN 9)

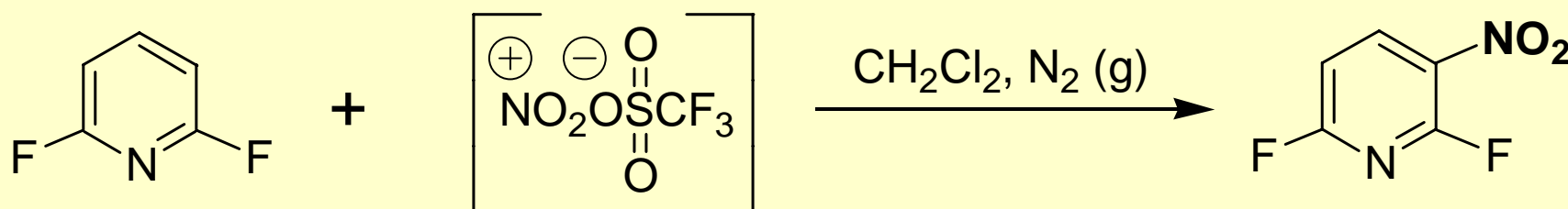




# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



## ➤ REFLUX TEMPERATURE NITRATION:



<u>CONDITIONS</u>	<u>NO<sub>2</sub>OSO<sub>2</sub>CF<sub>3</sub> EQUIV.</u>	<u>[REACTANT CONC.]</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
RT, 42 h	1.6	0.25M	69%	-----	<69%
RT, 24 h	1.5	1.09M	78%	98%	78%
REFLUX, 8 h	1.5	1.09M	100%	98%	99%*

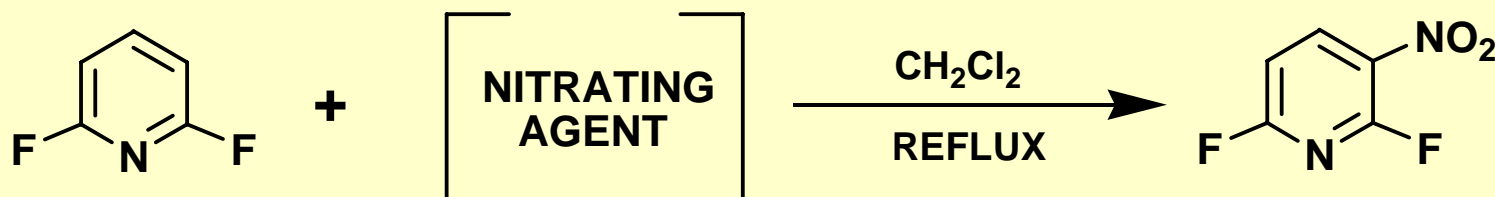
\* ANALYTICALLY PURE BY C, H, F, and N, ELEMENTAL ANALYSES



# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



## ➤ $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION EFFECTIVENESS:



	<u>TIME (h)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
$[\text{NO}_2\text{BF}_4^{\text{A}}]$	24	20%	-----
$[\text{NO}_2\text{OSO}_2\text{CF}_3]$	8	98%	99%*
*ANALYTICALLY PURE BY C, H, F, and N ELEMENTAL ANALYSES			
$[\text{HNO}_3 (100\%) / \text{H}_2\text{SO}_4^{\text{B}}]$	23	78% <sup>C</sup>	-----
<sup>B</sup> NEAT, 0 °C to RT		<sup>C</sup> DISTILLED YIELD	

A. Duffy, J. L.; Laali, K. K. *J. Org. Chem.* **1991**, 56, 3006-3009.

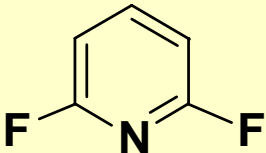
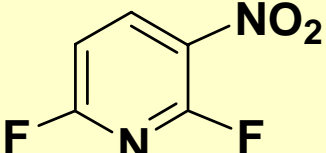
B. Mutterer, F.; Weiss, C. D. *Helv. Chem. Acta* **1976**, 59, 229-235.



# CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION

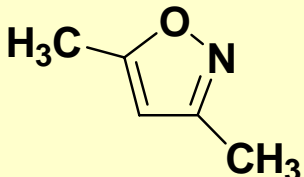



## ➤ REFLUX REACTION TIMES:

<u>REACTANT</u>	<u>TIME (h)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>	<u>PRODUCT</u>
	8*	98%	99%**	
	24	94%	99%**	

\* MINIMUM REACTION TIME

\*\* ANALYTICALLY PURE BY C, H, F, and N ELEMENTAL ANALYSES

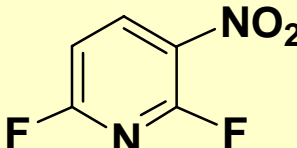
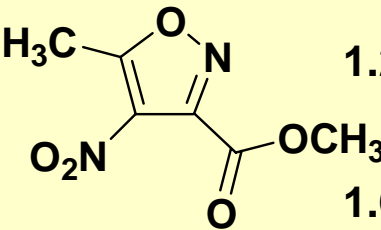
	3.5	92%	95%	
	16	90%	95%	



# CONVENTIONAL VS. MICROWAVE NO<sub>2</sub>OSO<sub>2</sub>CF<sub>3</sub> NITRATION COMPARISON



## ➤ COMPARATIVE PRODUCT RESULTS:

	<u>CONC.</u>	<u>CONDITIONS</u>	<u>SCALE (mmol)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
	1.2M	<b>PERSONAL MICROWAVE</b> 80 °C, 15 min.	3.4	94%	100%*
	1.3M	<b>MILESTONE MICROWAVE</b> 80 °C, 15 min.	54.5	98%	98%*
	1.1M	<b>CONVENTIONAL BENCHTOP</b> REFLUX, 8 h	16.4	98%	99%*
	1.2M	<b>PERSONAL MICROWAVE</b> 60 °C, 15 min.	3.4	100%	99%*
	1.6M	<b>MILESTONE MICROWAVE</b> 60 °C, 15 min.	27.0	100%	99%*
	UNREPORTED				
	1.2M	<b>CONVENTIONAL BENCHTOP</b> REFLUX, 24 h	17.2	100%	99%**

\* ANALYTICALLY PURE

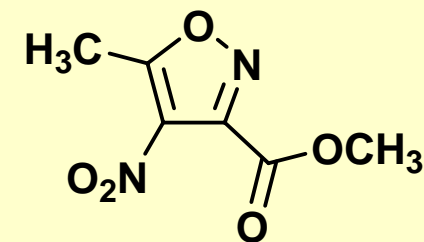
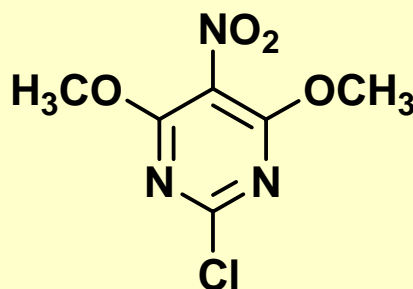
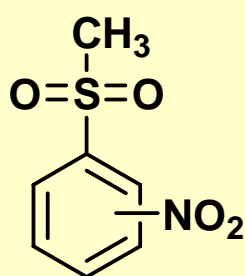
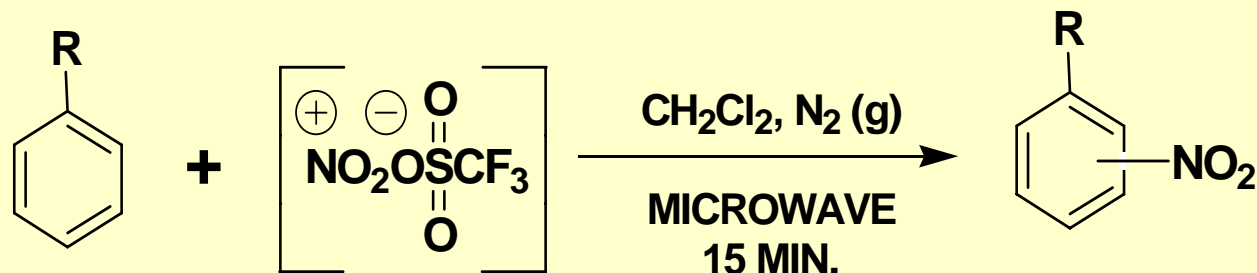
\*\* ELEMENTAL ANALYSIS NOT CONDUCTED



# MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



## ➤ MICROWAVE-ASSISTED PRODUCT RESULTS:



TEMP.°C)

80

60

60

60

ISOLATED  
YIELD

94%

91%

92%

100%

(O/M/P = 17/79/4)

PURITY

100%\*

100%\*\*

100%\*\*

100%\*\*

\* ISOMERIC MIXTURE ANALYTICALLY PURE

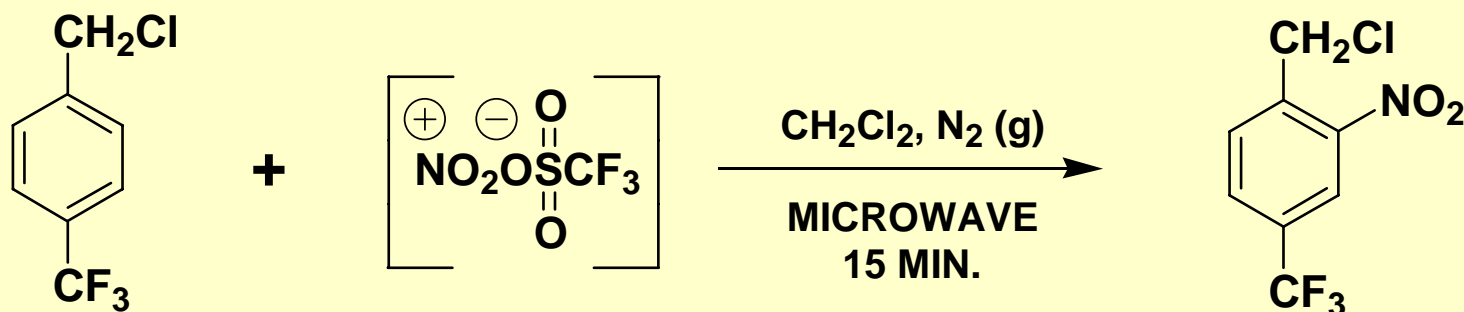
\*\* ANALYTICALLY PURE



# MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



## ➤ MICROWAVE-ASSISTED PRODUCT RESULTS:



<u>RUN NO.</u>	<u><math>\text{NO}_2\text{OSO}_2\text{CF}_3</math> EQUIV.</u>	<u>TEMP (°C)</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
1	1.5	80	82	-----	-----
2	1.5	100	82	-----	-----
3	2.0	80	100	98%	95%

- MINIMUM CONVERSION TEMPERATURE AT LEAST 80 °C (RUNS 1 & 2)
- $\text{NO}_2\text{OSO}_2\text{CF}_3$  EQUIVALENT INCREASE NEED FOR COMPLETE CONVERSION (RUN 3)



# MICROWAVE & CONVENTIONAL $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION

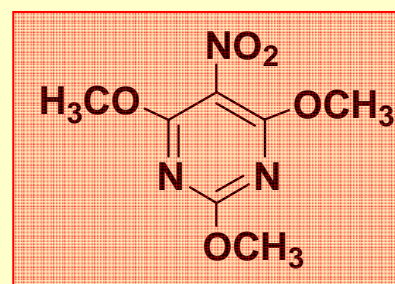
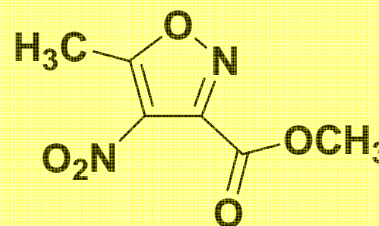
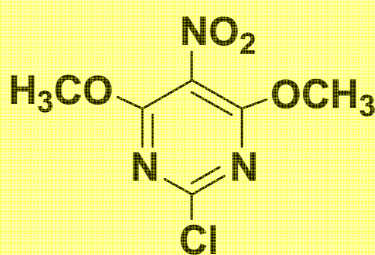
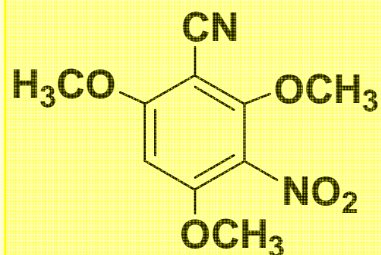


## ➤ $(\text{CH}_3)_4\text{NNO}_3$ BASED NITRONIUM TRIFLATE NITRATION:

### ▪ SIMPLE & CONVENIENT REACTION METHOD

- ✓ ONE-POT REACTION
- ✓ EASY AQUEOUS WORK-UP
- ✓ ISOMERIC SELECTIVITY
- ✓ EXCELLENT PRODUCT YIELDS
- ✓ HIGH (ANALYTICAL) "CRUDE" PRODUCT PURITY
- ✓ DIRECTLY SCALEABLE

### ▪ NEW COMPOUNDS & NEW SYNTHESIS ROUTE (DIRECT NITRATION)



Shackelford, S. A.; Anderson, M. B.; Christie, L. C.; Goetzen, T.; Guzman, M. C.; Hananel, M. A.; Kornreich, W. D.; Haitao, L.; Pathak, V. P.; Rabinovich, A. K.; Rajapakse, R. J.; Truesdale, L. K.; Tsank, S. M.; Vazir H. N.  
*J. Org. Chem.*, 2003, 68, 267-275.

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# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



## BIOTAGE MICROWAVE INITIATOR™



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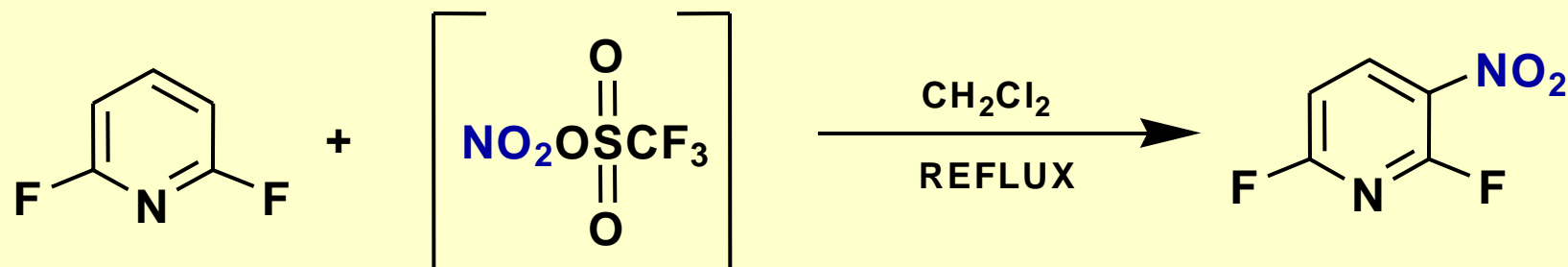




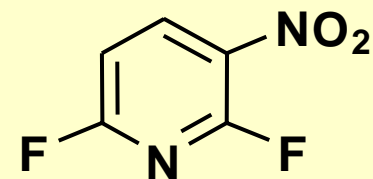
# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



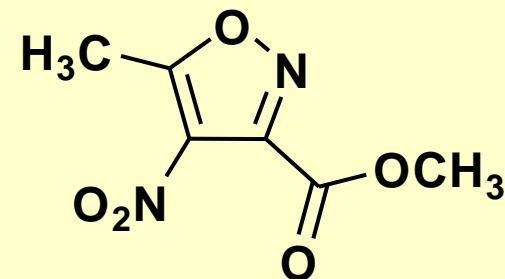
## ➤ MINIMUM NITRATION REACTION TIME:



<u>TEMP.</u>	<u>TIME (MIN)</u>	<u>PRODUCT CONVERSION</u>
<i>MW SET</i>	2	79%
= 80°C	3	98% (2 RUNS)
<i>ACTUAL</i>	4	100%
= 98-80°C		



<i>MW SET</i>	3	93%
= 60°C	5	100%
<i>ACTUAL</i>		
= 68-60°C		



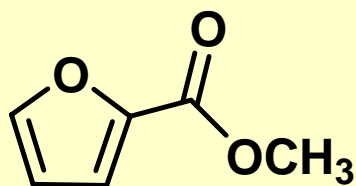
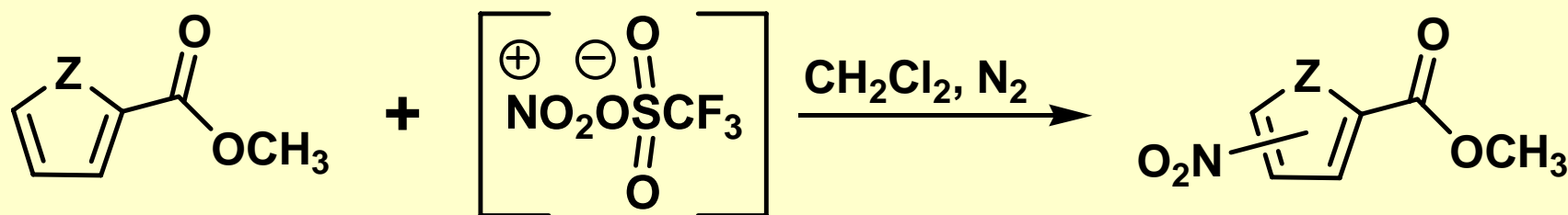


# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



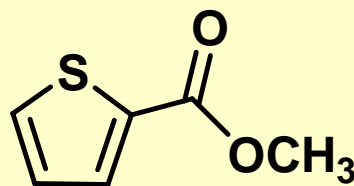
## ➤ 5-MEMBERED HETEROAROMATICS (ONE HETEROOATOM)

- DIFFERENT HETEROCYCLE REACTIVITY (Z = O, S, N)



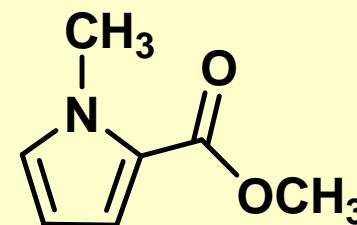
REACTIVE AT RT

NON-REACTIVE AT  
-78°C



-----

NON-REACTIVE AT  
-78°C



VERY REACTIVE AT RT

VERY REACTIVE AT  
-78°C



# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



## ■ HETERO-RING DIRECTING EFFECTS

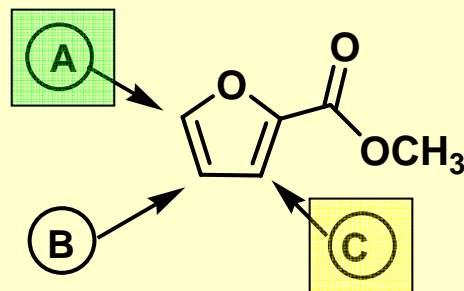
### CONDITIONS

RT to 60°C  
**MICROWAVE**  
1.1 M (5 MIN)

-78°C to 60°C  
**MICROWAVE**  
0.7M (5 MIN)

-78°C to RT  
**CONVENTIONAL**  
0.7M (3 HR)

-78°C to RT  
**CONVENTIONAL**  
0.09M  
(27-48 HRS)

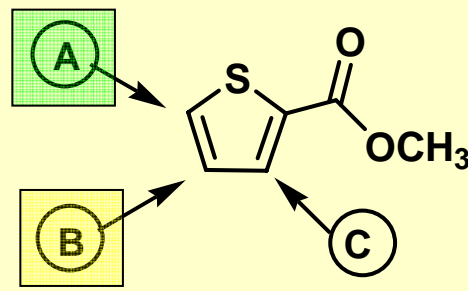


**A** **B** **C**  
95% 0% 5%

94% 0% 6%\*  
\*2% UNREACTED

97% 0% 3%

91-92% 0 9-8%

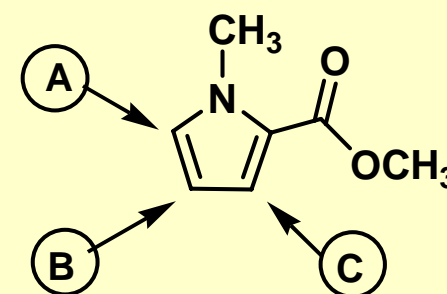


**A** **B** **C**  
---- ---- ----

59% 36% 5%(?)  
62% 38% ----

---- ---- ----

62% 38% 0%



NO NITRATED  
PRODUCT

**FURAN:** SLIGHT  
CONC. EFFECT

**THIOPHENE:**  
SLIGHT HEAT  
and./or CONC.  
EFFECT

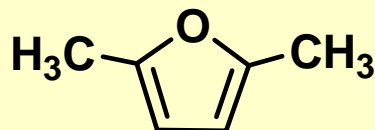
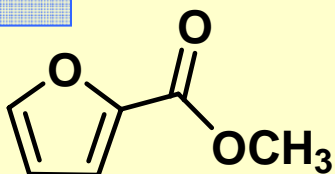


# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



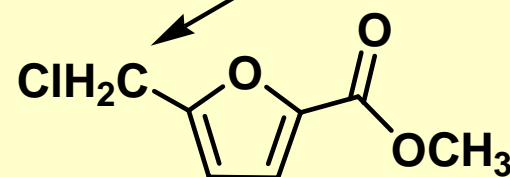
## ■ RING POSITIONAL REACTIVITY

SELECTIVE  
-NO<sub>2</sub> POSITION

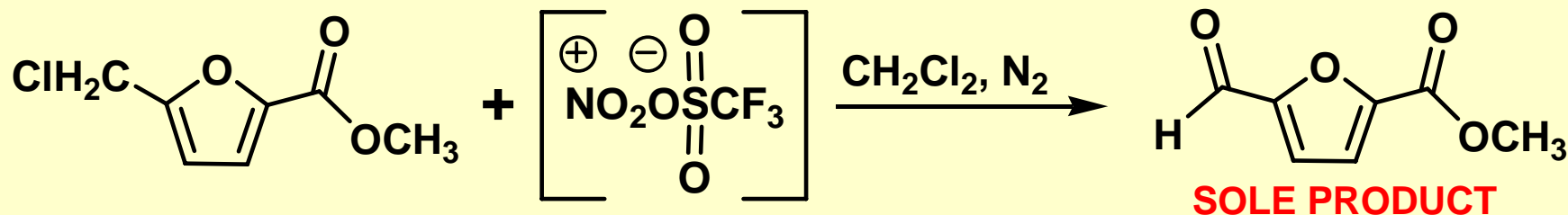


VERY REACTIVE AT  
RT & -78°C, NO PRODUCT

BENZYLIC-LIKE  
POSITION



NON-REACTIVE  
AT -78°C



**CONVENTIONAL** [0.19M, -78°C to RT, 76 HR]:

71% CONVERSION  
(NOT ISOLATED)

**MICROWAVE** [0.7M, -78°C to 70°C, 10 MIN]:

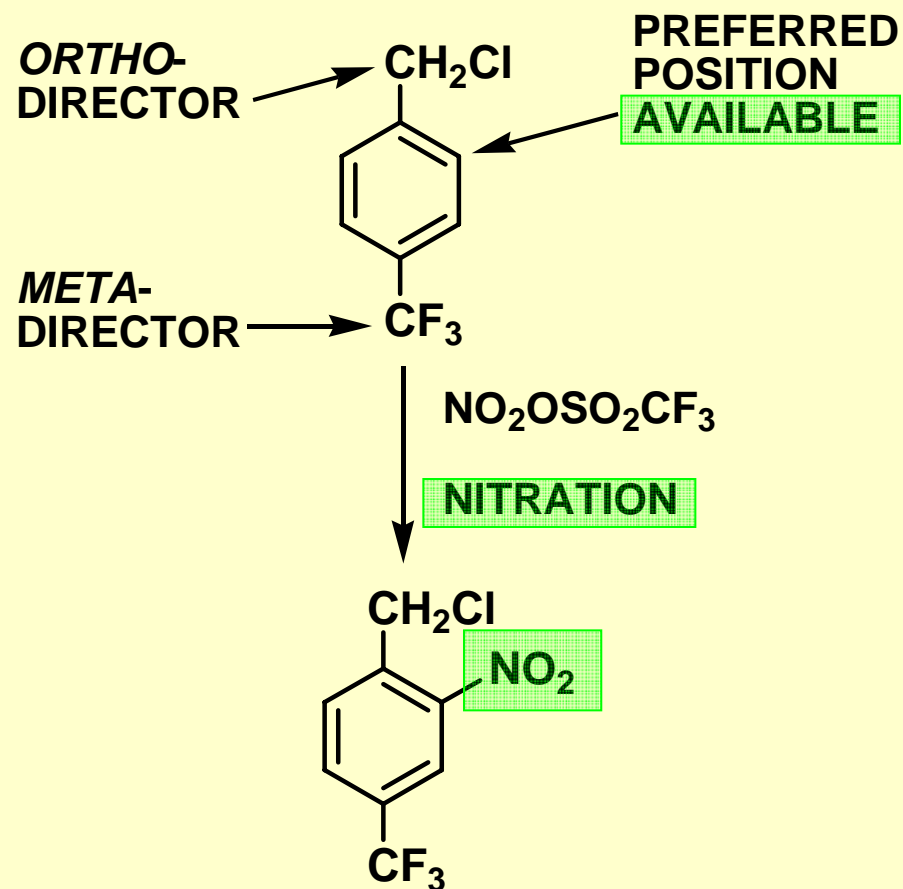
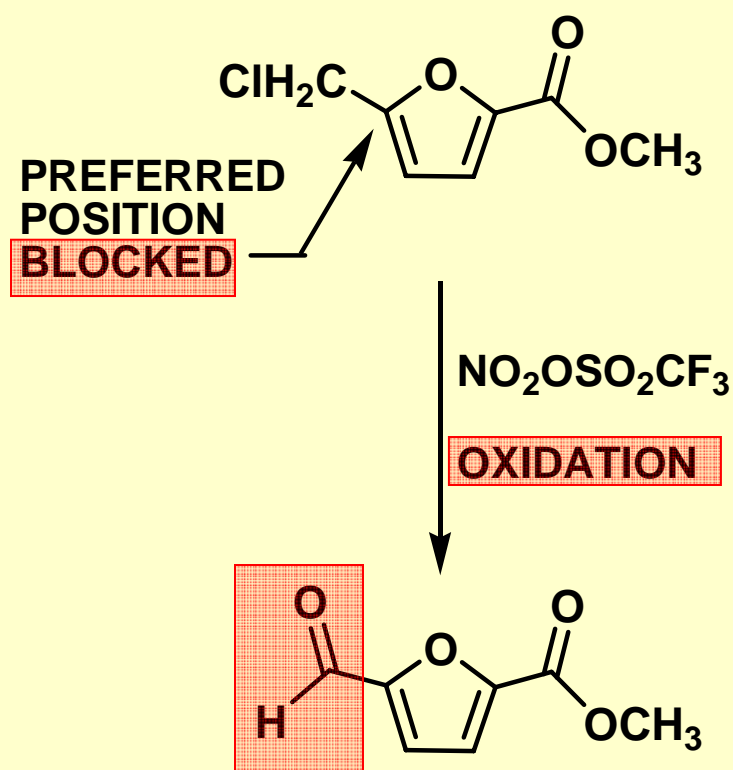
100% CONVERSION  
64% YIELD (NMR PURE)  
LT. YELLOW SOLID



# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



## ■ RING POSITION REACTIVITY



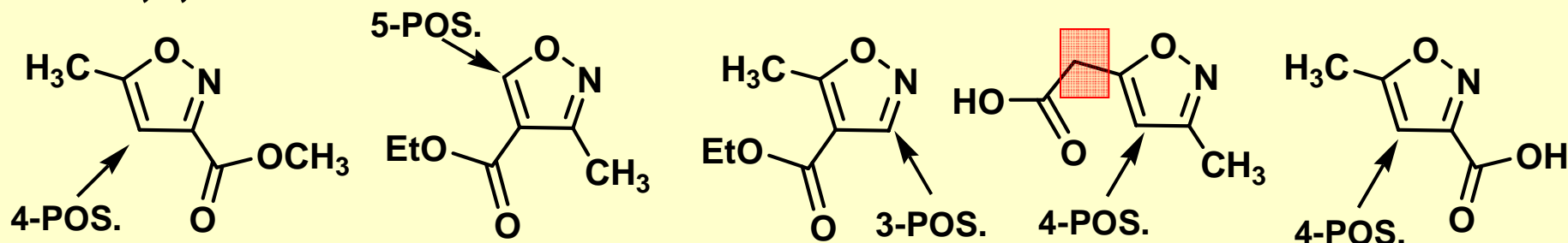


# MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



## ➤ 5-MEMBERED HETEROAROMATICS (TWO HETEROATOMS)

### ▪ 3,4,5-POSITION REACTIVITY PREFERENCE



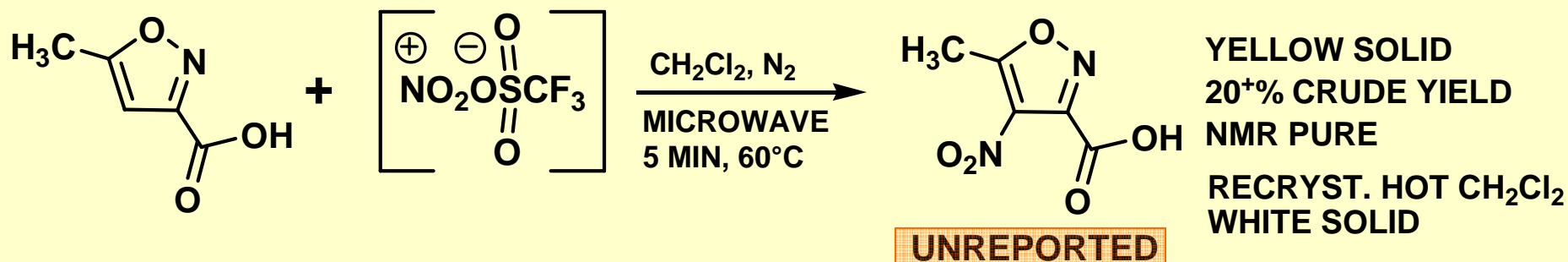
NON-REACTIVE  
AT RT  
HIGH YIELD  
NITRATION  
60°C, 5-15 MIN

NON-REACTIVE  
AT RT  
POSSIBLE TRACE  
NITRATION,  
60-100°C, 6-10 MIN

VERY REACTIVE  
AT RT  
NON-REACTIVE  
AT -78°C  
NO NITRATION

VERY REACTIVE  
AT -78°C  
NO NITRATION  
OBSERVED

SEE  
BELOW

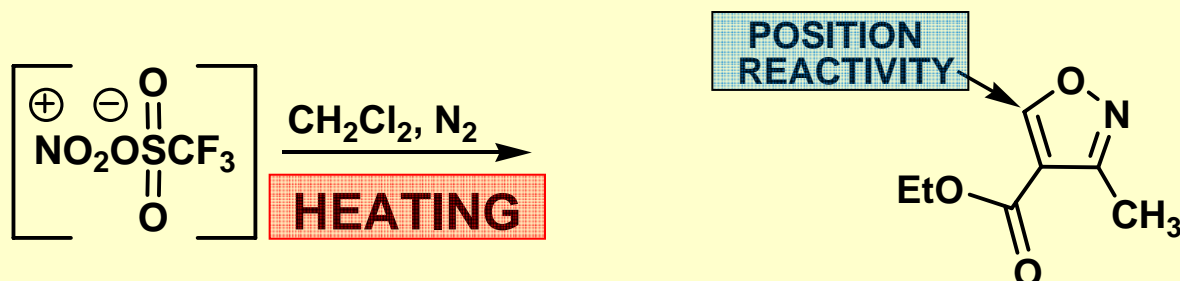




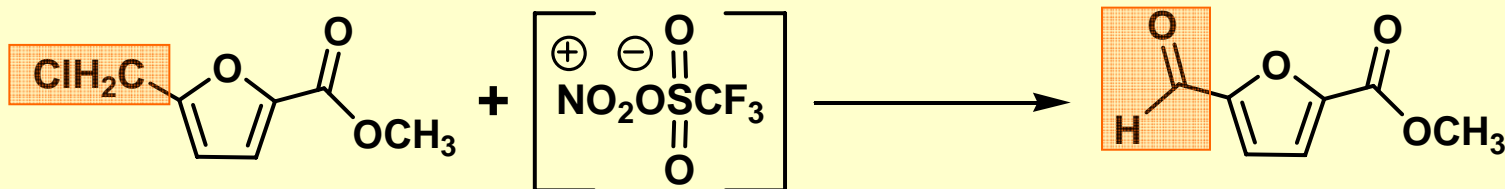
# SUMMARY



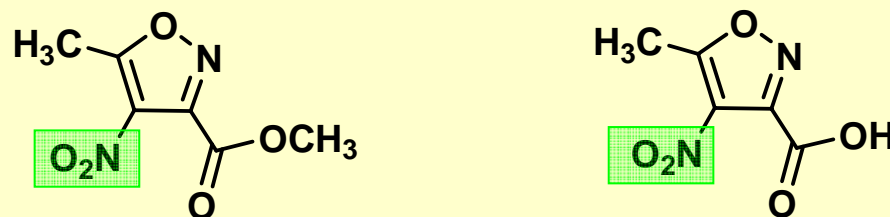
- **MICROWAVE-ASSISTED SYNTHESIS A USEFUL TOOL**
  - **SCREENING NITRONIUM TRIFLATE NITRATION SAFETY / REACTIVITY**



- **COMPLETING NEW NITRONIUM TRIFLATE REACTION PATHWAY**



- **OBTAINING NOVEL NITRO-HETEROAROMATICS w. NITRONIUM TRIFLATE**





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